Wolfe's mineral elixir (from medium 792)

$MgSO_4 \times 7\ H_2O$	30.00	g
$MnSO_4 \times H_2O$	5.00	g
NaCl	10.00	g
$FeSO_4 \times 7\ H_2O$	1.00	g
$CoCl_2 \times 6\ H_2O$	1.80	g
$CaCl_2 \times 2\ H_2O$	1.00	g
$ZnSO_4 \times 7\ H_2O$	1.80	g
$CuSO_4 \times 5\ H_2O$	0.10	g
$AlK(SO_4)_2 \times 12\ H_2O$	0.18	g
H_3BO_3	0.10	g
$Na_2MoO_4 \times 2\ H_2O$	0.10	g
$(NH_4)_2Ni(SO_4)_2 \times 6\ H_2O$	2.80	g
$Na_2WO_4 \times 2\ H_2O$	0.10	g
Na_2SeO_4	0.10	g
Distilled water	1000.00	ml

First adjust pH to 1.0 with diluted H_2SO_4, then add and dissolve the salts.

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