## SL-9

| | | |
|---|---|---|
| HCl (25%) | 10.00 | ml |
| $FeCl_2$ x 4 $H_2O$ | 2.00 | g |
| $ZnCl_2$ | 70.00 | mg |
| $MnCl_2$ x 4 $H_2O$ | 100.00 | mg |
| $H_3BO_3$ | 6.00 | mg |
| $CoCl_2$ x 6 $H_2O$ | 190.00 | mg |
| $CuCl_2$ x 2 $H_2O$ | 2.00 | mg |
| $NiCl_2$ x 6 $H_2O$ | 24.00 | mg |
| $Na_2MoO_4$ x 2 $H_2O$ | 36.00 | mg |
| Nitrilotriacetic acid | 12.80 | g |
| Distilled water | 990.00 | ml |

1. First dissolve $FeCl_2$ in the HCl, then dilute in water, add and dissolve the other salts. Finally make up to 1000.00 ml.

2. Adjust pH to 6.0 .

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