

Trace element solution

Nitrilotriacetic acid (NTA)	12.80	g
$\text{FeCl}_3 \times 6 \text{ H}_2\text{O}$	1.35	g
$\text{MnCl}_2 \times 4 \text{ H}_2\text{O}$	0.10	g
$\text{CoCl}_2 \times 6 \text{ H}_2\text{O}$	24.00	mg
$\text{CaCl}_2 \times 2 \text{ H}_2\text{O}$	0.10	g
ZnCl_2	0.10	g
$\text{CuCl}_2 \times 2 \text{ H}_2\text{O}$	25.00	mg
H_3BO_3	0.01	g
$\text{Na}_2\text{MoO}_4 \times 2 \text{ H}_2\text{O}$	24.00	mg
NaCl	1.00	g
$\text{NiCl}_2 \times 6 \text{ H}_2\text{O}$	0.12	g
Na_2SeO_4	4.00	mg
Na_2WO_4	4.00	mg
Distilled water	1000.00	ml

First dissolve the nitrilotriacetic acid and adjust pH to 6.5 with NaOH, and then add minerals. Final pH 7.0.