

Trace element solution

Nitrilotriacetic acid	12.800	g
$\text{FeCl}_3 \times 6 \text{ H}_2\text{O}$	1.350	g
$\text{MnSO}_4 \times \text{H}_2\text{O}$	0.100	g
$\text{CoCl}_2 \times 6 \text{ H}_2\text{O}$	0.024	g
$\text{CaCl}_2 \times 2 \text{ H}_2\text{O}$	0.100	g
ZnCl_2	0.100	g
$\text{CuCl}_2 \times 5 \text{ H}_2\text{O}$	0.025	g
H_3BO_3	0.010	g
$\text{Na}_2\text{MoO}_4 \times 2 \text{ H}_2\text{O}$	0.024	g
NaCl	1.000	g
$\text{NiCl}_2 \times 6 \text{ H}_2\text{O}$	0.120	g
$\text{Na}_2\text{SeO}_3 \times 5 \text{ H}_2\text{O}$	4.000	mg
Na_2WO_4	4.000	mg
Distilled water	1000.000	ml

First dissolve nitrilotriacetic acid and adjust pH to 6.5 with KOH, then add minerals. Final pH 7.0 (with KOH).