



Solution II

KH_2PO_4	3.00	g
$(\text{NH}_4)_2\text{SO}_4$	6.00	g
NaCl	6.00	g
$\text{MgSO}_4 \times 7 \text{ H}_2\text{O}$	0.60	g
$\text{CaCl}_2 \times 2 \text{ H}_2\text{O}$	0.60	g
Distilled water	1000.00	ml