

**Modified Wolin's mineral solution II** (from medium 700)

Nitrolotriacetic acid	1.500	g
MgSO <sub>4</sub> x 7 H <sub>2</sub> O	3.000	g
MnSO <sub>4</sub> x H <sub>2</sub> O	0.500	g
NaCl	1.000	g
FeSO <sub>4</sub> x 7 H <sub>2</sub> O	0.100	g
CoSO <sub>4</sub> x 7 H <sub>2</sub> O	0.180	g
CaCl <sub>2</sub> x 2 H <sub>2</sub> O	0.100	g
ZnSO <sub>4</sub> x 7 H <sub>2</sub> O	0.180	g
CuSO <sub>4</sub> x 5 H <sub>2</sub> O	0.010	g
AlK(SO <sub>4</sub> ) <sub>2</sub> x 12 H <sub>2</sub> O	0.020	g
H <sub>3</sub> BO <sub>3</sub>	0.010	g
Na <sub>2</sub> MoO <sub>4</sub> x 2 H <sub>2</sub> O	0.010	g
NiCl <sub>2</sub> x 6 H <sub>2</sub> O	0.025	g
Na <sub>2</sub> SeO <sub>3</sub> x 5 H <sub>2</sub> O	0.300	mg
Distilled water	1000.000	ml

First dissolve nitrolotriacetic acid and adjust pH to 6.5 with KOH, then add minerals. Adjust final to pH 7.0 with KOH.