

1315: ANAEROBIC THAUERA MEDIUM

This recipe contains strain-specific modifications for Acidovorax sp. DSM 15015 *

Final pH: 7.1 - 7.5 Final volume: 1000 ml

KNO ₃	0.85	g
NaHCO ₃	2.52	g
CaCl ₂	0.01	g
$MgSO_4 \times 7 H_2O$	0.50	g
MnSO ₄	0.01	g
NH ₄ Cl	0.30	g
NaCl	0.05	g
Selenite and tungstate solution	2.00	ml
Distilled water	884.00	ml
Phosphate solution	100.00	ml
Acetate solution 98.0 g/L	10.00	ml
Vitamin solution	5.00	ml
Trace element solution SL-10	1.00	ml
3,4-dihydroxybenzoate	0.77	g/l

Distribute to gas tight vessels, gas with 90% N_2 +10% CO₂ to achieve anaerobic conditions and autoclave. After cooling, add the phosphate solution, substrate solution, vitamin solution and trace element solution. After mixing, pH should be 7.1 - 7.5.

* Replace fatty acid by 3,4-dihydroxybenzoate (0.77 g/l)

1.45	g
0.25	g
100.00	ml
	1.45 0.25 100.00

Distribute to gas tight vessels under N_2 , autoclave.

Acetate solution 98.0 g/L		
K-acetate	9.80	g
Distilled water	100.00	ml

Filter sterilize and gas with N_2 . For strains <u>DSM 12141</u>, <u>DSM 12142</u>, <u>DSM 11243</u> and <u>DSM 12144</u>, acetate may be replaced by Na-capronate (stock solution 2.76 g/100 ml).

Microorganisms

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Vitamin solution (from medium 461)			
Vitamin B ₁₂	50.00	mg	
Pantothenic acid	50.00	mg	
Riboflavin	50.00	mg	
Pyridoxamine hydrochloride	10.00	mg	
Biotin	20.00	mg	
Folic acid	20.00	mg	
Nicotinic acid	25.00	mg	
Nicotine amide	25.00	mg	
	50.00	- mg	_
p-Aminobenzoic acid	50.00	mg	
Thiamine-HCl x 2 H_2O	50.00	mg	
Distilled water	1000.00	ml	

Stir for some hours, filter sterilize the solution.

Trace element solution	SL-10 (from medium 320)
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HCI (25%)	10.00	ml
$FeCl_2 \times 4 H_2O$	1.50	g
ZnCl ₂	70.00	mg
$MnCl_2 \times 4 H_2O$	100.00	mg
H ₃ BO ₃	6.00	mg
$CoCl_2 \times 6 H_2O$	190.00	mg
$CuCl_2 \times 2 H_2O$	2.00	mg
$NiCl_2 \times 6 H_2O$	24.00	mg
$Na_2MoO_4 \times 2 H_2O$	36.00	mg
Distilled water	990.00	ml

First dissolve $FeCl_2$ in the HCl, then dilute in water, add and dissolve the other salts. Finally make up to 1000.00 ml.

Selenite and tungstate solution		
Distilled water	1000.00	ml
NaOH	0.50	g
$Na_2SeO_3 \times 5 H_2O$	3.00	mg
$Na_2WO_4 \ge H_2O$	4.00	mg

First dissolve NaOH, subsequently the metal salts. Distribute to gas tight vessels under $N_{\rm 2},$ autoclave.